

# Course guide 220126 - TM - Thermodynamics of Materials

Last modified: 19/04/2023

Unit in charge:	Terrassa School of Industrial, Aerospace and Audiovisual Engineering
Teaching unit:	724 - MMT - Department of Heat Engines.
Degree:	BACHELOR'S DEGREE IN AEROSPACE TECHNOLOGY ENGINEERING (Syllabus 2010). (Optional subject). BACHELOR'S DEGREE IN AEROSPACE VEHICLE ENGINEERING (Syllabus 2010). (Optional subject). BACHELOR'S DEGREE IN INDUSTRIAL TECHNOLOGY ENGINEERING (Syllabus 2010). (Optional subject).

Academic year: 2023 ECTS Credits: 3.0 Languages: English

### **LECTURER**

Coordinating lecturer:	Calventus Sole, Yolanda
Others:	Roman Concha, Frida Rosario

### **TEACHING METHODOLOGY**

The course is divided into parts:

Theory classes

Practical classes (Laboratory Sessions)

Self-study for doing exercises and activities.

In the theory classes, teachers will introduce the theoretical basis of the concepts, methods and results and illustrate them with appropriate examples to facilitate their understanding.

The practical classes will take place in the Laboratory, and in them, students will observe the different phenomena presented in the theory classes.

Students need to work independently on the materials provided by teachers in order to assimilate the concepts. The teachers control activities by ATENEA.

# LEARNING OBJECTIVES OF THE SUBJECT

- Analyse the thermal behavior of materials, specially polymeric materials. Glass transition phenomenon, crystallisation process, melting process, curing process.

- Main characteristics of other phenomena such as superconductivity

- Learn how to handle some of the experimental techniques of thermal analysis used in industry (pharmaceutical, electronics...) and in laboratories/research centers.

### **STUDY LOAD**

Туре	Hours	Percentage
Hours large group	30,0	40.00
Self study	45,0	60.00

Total learning time: 75 h



### CONTENTS

### Module 1: First and Second order phase transition

### **Description:**

Study and analyze the thermal transitions that occur in materials and the consequences they have on their structure, properties and technical applications.

It is seen experimentally using the DSC technique how these phase changes occur in practise

#### Specific objectives:

Know how to identify the technical transitions that appear in a material and what they mean in terms of its structure and properties. What technical advantages can these phase changes bring us?

Begin to develop in the use of the technique of Differential Scanning Calorimetry (DSC)

### **Related activities:**

Activity 1: Theory and Problem Classes Activity 2: Problem solving Activity 2: Find the foundamentals and the main applications of a certain thermal transition Activity 3: Laboratory Session 1

**Full-or-part-time:** 15h Theory classes: 8h

Self study : 7h

#### Module 2: Polymer overview

#### **Description:**

It gives an overview of what a polymer is, what kind of polymers we have, its main characteristics and properties. Industrial applications and in daily life

### **Specific objectives:**

- What is a polymer
- What is a thermoplastic, a thermostable and an elastomer
- How they are formed and how they behave
- How they can be identified
- Know its properties

**Related activities:** 

Activity 4. (The presentation will be done at the end of the semester)

Full-or-part-time: 12h Theory classes: 2h Self study : 10h



### Module 3: Crystallisation process in polymers

### **Description:**

- Introduction and concepts of amourphous and crystalline structure
- Crystallisation kinetics. Nucleation and Growth
- The crystallization morphology: spherulites
- Factors which affect the crystallisation process
- Properties related to the crystalline structure
- Melting temperatures, enthalpies and entropies of melting
- Crystallisation study using DSC technic

### Specific objectives:

Know how to distinguish between amorphous phase and crystalline phase Know that polymers can be semicrystalline or amorphous (never fully crystallines) Know how to interpret in a DSC curve the different thermal transitions in thermoplastic polymers Know how to calculate the crystallized fraction and the degree of crystallinity using DSC results Introduce the use of experimental technic TGA (Thermogravimetric Analysis)

### **Related activities:**

Activity 1: Theory classes Activity 3: Laboratory Session 2 Activity 3: Laboratory Session 3

### Full-or-part-time: 18h

Theory classes: 6h Self study : 12h

### Module 4: Glass transition phenomenon

### **Description:**

### Introduction

Factos affecting glass transition temperature How to obtai glass transition temperature Changes in physical properties Pseudo second order thermal transition Phenomenology Fictive Temprature Physical Aging

### **Specific objectives:**

Concept of glass transition Concept of vitreous state - rubbery state Concept of physical aging

### **Related activities:**

Activity 2: Solve an Athenea Quizz in class Activity 2: Class activity on physical aging Activity 4: Laboratory session 4

Full-or-part-time: 15h Theory classes: 7h Self study : 8h



### Module 5: Curing reaction thermosetting polymers

### **Description:**

- What is a thermoset
- How to produce a thermoset:
- Main concepts:
- Curing process
- Gelation
- Time of gel
- Vitrification

**Specific objectives:** Know the main characteristics, properties and applications of thermosets

**Related activities:** Activity 2: Make a search about thermosets applications Activity 4: Laboratory session 5

**Full-or-part-time:** 15h Theory classes: 7h Self study : 8h

### ACTIVITIES

#### **Activity 1: Theory and Problem Classes**

**Description:** Presentation of the content using audiovisual media

#### **Specific objectives:**

Explain the contents of the subject and learn to relate the concepts to be able to solve practical problems

**Material:** Computer Blackboard Video

Full-or-part-time: 18h Theory classes: 18h

### **Activity 2: Continous Assessment Activities**

### **Description:**

- Solving problems.
- Solving Athenea quizzes
- Do searches to delve into certain concepts in the syllabus

### Specific objectives:

Do tasks and quizzes that help to achieve the contents of the subject

Material: Computer

**Delivery:** Via Athenea

**Full-or-part-time:** 15h Self study: 15h



### **Activity 3: Laboratory sessions**

### **Description:**

Laboratory sessions

#### Specific objectives:

Learn how to use the DSC and TGA. These techniques are widely used in industry and in research centers and laboratories.

Learn to identify correctly the meaning of the peaks obtained in the curves.

Perform the necessary numerical calculations, based on the experimental data obtained, in order to obtain the values of the properties needed to understand the behavior of the material

#### Material:

Lab guide Differential scanning calorimeter (DSC) Thermogravimetric analizer (TGA) Computer Samples of different materials in order to carry out different experiments and studies: Indium, zinc and different polymeric materials

### **Delivery:** In groups of 2 students via Athenea

**Full-or-part-time:** 30h Theory classes: 10h Self study: 20h

#### Activity 4: Present an essay about a polymeric material studied in the laboratory

### **Description:**

Carry out a essay about the manufacturing process of a given polymer, its chain structure, its properties and its applications.

The polymer will be chosen from those studied in the laboratory

It will be done in groups of 2 students

It will be presented in the last class. The presentation should be about 15 minutes per group.

### **Specific objectives:**

Do an activity that includes everything that has been worked in the subject

## Material:

Computer

### Delivery:

Upload in Athenea the presentation file (pdf or ppt)

### **Full-or-part-time:** 12h Theory classes: 2h Self study: 10h



### **GRADING SYSTEM**

The final grade depends on the following criteria:

- Continous assessment: 25%
- Write and present an essay about a polymer studied in laboratory: 25%
- Laboratory Reports: 50%

### **BIBLIOGRAPHY**

#### **Basic:**

- Adkins, C. J. Equilibrium thermodynamics. 3rd ed. Cambridge [etc.]: Cambridge University Press, 1983. ISBN 0521254450.

- Billmeyer, Fred W. Textbook of polymer science. 3rd ed. New York: Wiley-Interscience. Division of John Wiley & Sons, 1984. ISBN 0471828343.

- Smith J. M. [et al.]. Introduction to chemical engineering thermodynamics. 8th ed. New York: McGraw-Hill, 2018. ISBN 9781259696527.

- Mandelkern, L. Crystallization of polymers. Vol. 1, Equilibrium concepts. 2nd ed. Cambridge: Cambridge University Press, 2010. ISBN 9780521020138.

### **Complementary:**

- Dusek, K. Epoxy resins and composites. Springer-Verlag, 2013. ISBN 9783662159644.