



## Course guide

### 340022 - QUIM-N1O13 - Chemistry

Last modified: 28/06/2023

**Unit in charge:** Vilanova i la Geltrú School of Engineering  
**Teaching unit:** 713 - EQ - Department of Chemical Engineering.

**Degree:** BACHELOR'S DEGREE IN ELECTRICAL ENGINEERING (Syllabus 2009). (Compulsory subject).  
BACHELOR'S DEGREE IN INDUSTRIAL DESIGN AND PRODUCT DEVELOPMENT ENGINEERING (Syllabus 2009). (Compulsory subject).  
BACHELOR'S DEGREE IN INDUSTRIAL ELECTRONICS AND AUTOMATIC CONTROL ENGINEERING (Syllabus 2009). (Compulsory subject).  
BACHELOR'S DEGREE IN MECHANICAL ENGINEERING (Syllabus 2009). (Compulsory subject).

**Academic year:** 2023    **ECTS Credits:** 6.0    **Languages:** Catalan

#### LECTURER

**Coordinating lecturer:** NATIVITAT SALVADÓ CABRÉ

**Others:** SALVADOR BUTÍ PAPIOL  
AGUSTÍ FORTUNY SANROMÀ  
NÚRIA JIMÉNEZ GARCÍA  
JOAQUIM OLIVÉ DURAN  
LURDES ROSET CALZADA  
MONTSERRAT RUIZ PLANAS  
NATIVITAT SALVADÓ CABRÉ

#### PRIOR SKILLS

Students who have not done Chemistry subjects in the higher school or in the higher degree modules, a detailed reading of some chemistry text for the higher school is recommended, also a reading of the pages, F1 to F104, of "FUNDAMENTOS" that you will find in the book Principles of Chemistry. This book is recommended in the bibliography and can be found in the EPSEVG library:

Authors: Peter Atkins / Loretta Jones  
Principios de Química. Los caminos del descubrimiento  
5a Edición. Editorial Médica Panamericana

These pages contain an introduction to the chemical language, the most basic concepts, some examples and exercises.

#### DEGREE COMPETENCES TO WHICH THE SUBJECT CONTRIBUTES

##### Specific:

5. CE4. Ability to understand and apply principles of basic knowledge of general chemistry, organic and inorganic chemistry and its applications in engineering.

##### Transversal:

1. TEAMWORK - Level 1. Working in a team and making positive contributions once the aims and group and individual responsibilities have been defined. Reaching joint decisions on the strategy to be followed.
2. SELF-DIRECTED LEARNING - Level 1. Completing set tasks within established deadlines. Working with recommended information sources according to the guidelines set by lecturers.
3. EFFICIENT ORAL AND WRITTEN COMMUNICATION - Level 1. Planning oral communication, answering questions properly and writing straightforward texts that are spelt correctly and are grammatically coherent.
4. SUSTAINABILITY AND SOCIAL COMMITMENT - Level 1. Analyzing the world's situation critically and systemically, while taking an interdisciplinary approach to sustainability and adhering to the principles of sustainable human development. Recognizing the social and environmental implications of a particular professional activity.



## TEACHING METHODOLOGY

- 1 Teaching in the classroom
2. Use of teaching material: visualizations, animations and videos
3. Solving exercises and problems, applications of chemistry to the everyday world.
4. Experimental chemistry in the laboratory
5. Student-teacher interaction based on exercises and questions proposed in classroom
- 6 Teacher-student interaction during time scheduled

## LEARNING OBJECTIVES OF THE SUBJECT

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## STUDY LOAD

Type	Hours	Percentage
Self study	90,0	60.00
Hours large group	30,0	20.00
Hours small group	30,0	20.00

**Total learning time:** 150 h

## CONTENTS

### Structure of matter

#### Description:

- Introduction to chemistry
- Electronic structure
- Periodic Table and Periodic Properties
- Chemical bonds
- Molecular structure
- States of matter: gas, solid and liquid
- Solutions

#### Related activities:

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**Full-or-part-time:** 60h

Theory classes: 12h

Laboratory classes: 12h

Self study : 36h



## Chemical Transformations

### Description:

- Chemical reactions
- Thermochemistry
- Chemical Kinetics
- Chemical equilibrium
- Acid and based
- Reduction-oxidation reactions, electrochemistry

### Full-or-part-time: 90h

Theory classes: 18h

Laboratory classes: 18h

Self study : 54h

## GRADING SYSTEM

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## BIBLIOGRAPHY

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### Basic:

- Chang, Raymond; Overby, Jason. Química [on line]. 13a ed. México: McGraw-Hill, 2020 [Consultation: 15/02/2024]. Available on: [https://www-ingebook-com.recursos.biblioteca.upc.edu/ib/NPcd/IB\\_BooksVis?cod\\_primaria=1000187&codigo\\_libro=10619](https://www-ingebook-com.recursos.biblioteca.upc.edu/ib/NPcd/IB_BooksVis?cod_primaria=1000187&codigo_libro=10619). ISBN 9781456277161.
- Atkins, P. W. Principios de química: los caminos del descubrimiento. 5a ed. Buenos Aires [etc.]: Médica Panamericana, 2012. ISBN 9789500602822.

### Complementary:

- Gray, Theodore W. The Elements : a visual exploration of every known atom in the universe. New York: Black Dog & Leventhal, 2009. ISBN 9781579128142.
- Chang, Raymond. Fundamentos de Química [on line]. Madrid: McGraw-Hill, 2011 [Consultation: 07/03/2024]. Available on: [https://www-ingebook-com.recursos.biblioteca.upc.edu/ib/NPcd/IB\\_BooksVis?cod\\_primaria=1000187&codigo\\_libro=5646](https://www-ingebook-com.recursos.biblioteca.upc.edu/ib/NPcd/IB_BooksVis?cod_primaria=1000187&codigo_libro=5646). ISBN 9786071505415.

## RESOURCES

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### Audiovisual material:

- Nom recurs. Resource

### Other resources:

<https://www.rsc.org/periodic-table>

/> <https://www.euchems.eu/wp-content/uploads/2018/10/CATALAN-Periodic-Table-Element-Scarcity.pdf> />