



## Course guides

# 205226 - 205226 - Surface Chemistry for Industrial Applications Design

Last modified: 29/05/2020

**Unit in charge:** Terrassa School of Industrial, Aerospace and Audiovisual Engineering  
**Teaching unit:** 713 - EQ - Department of Chemical Engineering.

**Degree:** BACHELOR'S DEGREE IN INDUSTRIAL TECHNOLOGY ENGINEERING (Syllabus 2010). (Optional subject).  
BACHELOR'S DEGREE IN AEROSPACE TECHNOLOGY ENGINEERING (Syllabus 2010). (Optional subject).  
BACHELOR'S DEGREE IN AEROSPACE VEHICLE ENGINEERING (Syllabus 2010). (Optional subject).  
BACHELOR'S DEGREE IN INDUSTRIAL ELECTRONICS AND AUTOMATIC CONTROL ENGINEERING (Syllabus 2009). (Optional subject).  
BACHELOR'S DEGREE IN ELECTRICAL ENGINEERING (Syllabus 2009). (Optional subject).  
BACHELOR'S DEGREE IN MECHANICAL ENGINEERING (Syllabus 2009). (Optional subject).  
BACHELOR'S DEGREE IN CHEMICAL ENGINEERING (Syllabus 2009). (Optional subject).  
BACHELOR'S DEGREE IN AUDIOVISUAL SYSTEMS ENGINEERING (Syllabus 2009). (Optional subject).  
BACHELOR'S DEGREE IN TEXTILE TECHNOLOGY AND DESIGN ENGINEERING (Syllabus 2009). (Optional subject).  
BACHELOR'S DEGREE IN INDUSTRIAL DESIGN AND PRODUCT DEVELOPMENT ENGINEERING (Syllabus 2010). (Optional subject).

**Academic year:** 2020    **ECTS Credits:** 3.0    **Languages:** English

## LECTURER

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**Coordinating lecturer:** MANUEL-JOSE LIS ARIAS

**Others:**

## PRIOR SKILLS

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Multidisciplinary approach of the subject makes that the competences acquired until third curs should be enough. Advanced studies in third curs and totally passed the second course would be recommended. This condition is not a way to exclude students.

## TEACHING METHODOLOGY

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Some activities that correspond to the six groups of items in which subject has been divided. After summarized theoretical approach to each group, there will be some real problems to be solved

## LEARNING OBJECTIVES OF THE SUBJECT

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Real industrial systems require, for their modification or for new design of a transversal vision about the phenomena involved in the whole process. The majority of industrial applications involve heterogeneous systems that act just depending on the chemical characteristics of the components in contact.

The main objective of this subject is to define every phenomenology involved in this type of industrial systems



## STUDY LOAD

Type	Hours	Percentage
Hours large group	14,0	18.67
Self study	45,0	60.00
Hours medium group	8,0	10.67
Hours small group	8,0	10.67

**Total learning time:** 75 h

## CONTENTS

### Interfaces in industrial systems

**Description:**

Chemical characteristics of heterogeneous systems. Electric potential and diffusional layers. Colloidal systems. Amphoteric molecules and surface activity molecules: micelles, mixed micelles, biological membranes  
Relationship between chemical structure and properties depending on (viscosity, dipolar moment,...) Interfacial tension. Couy-Chapman Theory for different interfaces

**Specific objectives:**

To establish the relationship between chemical structure and surface properties

**Related activities:**

Case 1: For an specific system, detect and study the relationship between interfaces

**Full-or-part-time:** 12h 30m

Theory classes: 1h

Practical classes: 2h

Laboratory classes: 2h

Self study : 7h 30m

### Gas-liquid

**Description:**

Bubbles. Interfaces generated by the formation of bubbles. Foams. Final density control for foams. Antifoaming agents

**Specific objectives:**

To establish the basic phenomena involved in the contact between bubbles and liquids

**Full-or-part-time:** 12h 30m

Theory classes: 1h

Practical classes: 2h

Laboratory classes: 2h

Self study : 7h 30m



### Non-miscible liquid systems

**Description:**

Emulsions and dispersions. Continuous and dispersed phases. Characteristics of emulsifying and dispersing systems. Mechanisms involved in the formation of emulsions and dispersions. Applications: Cosmetics, coating, finishing, detergents.

**Specific objectives:**

To become familiar with two non-miscible liquids and know how to manage to mix them

**Related activities:**

Case 2: For an specific case, to establish and define the methodology to follow to make a disperse system

**Full-or-part-time:** 11h 30m

Theory classes: 4h

Self study : 7h 30m

### Non porous solids-liquids

**Description:**

Adhesion, concepts involved in the adhesion between non-porous solids . Solid-liquid interfacial tension. Wetting phenomena. Langmuir-Blodgett systems. Diffusional limiting layer. Vapour phase deposition. Surface modification with enzymatic treatments

**Specific objectives:**

To define the chemical interactions between non porous solids and liquids

**Full-or-part-time:** 11h 30m

Theory classes: 4h

Self study : 7h 30m

### Porous solids-liquids

**Description:**

Permeability, porosity. Capillarity effect. Porous diffusion. Catalytic systems in the adhesion in porous surfaces. Swelling effect in polymers. Chemical dependence between absorbate and absorbent.

**Specific objectives:**

To differenciate the behaviour of the solid interface when the solid is porous

**Related activities:**

Case 3: For specific cases, define the strategy to improve the adhesion phenomena

**Full-or-part-time:** 13h 30m

Theory classes: 2h

Practical classes: 2h

Laboratory classes: 2h

Self study : 7h 30m



## Gas-solid systems

### Description:

Porosity effects on solid-gas systems. Membrane effect. Filtration. Retention capability of gases by membranes, depending on their chemical characteristics.

### Specific objectives:

To establish the interactions between solid and gases

### Related activities:

Case 4: For an specific system, define and determine the interactions between solid and gas

### Full-or-part-time: 13h 30m

Theory classes: 2h

Practical classes: 2h

Laboratory classes: 2h

Self study : 7h 30m

## GRADING SYSTEM

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Developed that will weigh 25% each. Written delivery and oral presentation

## EXAMINATION RULES.

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The written report for each case and the oral presentations will be scheduled

## BIBLIOGRAPHY

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### Basic:

- Grumezescu, Alexandru. Surface chemistry of nanobiomaterials: applications of nanobiomaterials. New York: William Andrew, 2016. ISBN 9780323428613.
- Hiemenz, Paul C. Principles of colloid and surface chemistry. 3th. Boca Raton: CRC Press, 1997. ISBN 9780824793975.

## RESOURCES

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### Other resources:

For every case, some articles and specific bibliography will be delivered