Course guides
295024 - TERM - Thermodynamics

Unit in charge: Barcelona East School of Engineering
Teaching unit: 748 - FIS - Department of Physics.

Degree: BACHELOR’S DEGREE IN CHEMICAL ENGINEERING (Syllabus 2009). (Compulsory subject).
BACHELOR’S DEGREE IN MATERIALS ENGINEERING (Syllabus 2010). (Compulsory subject).

Academic year: 2021 ECTS Credits: 6.0 Languages: Catalan

LECTURER

Coordinating lecturer: JOSE LUIS TAMARIT MUR

Others: Primer quadrimestre:
MARIA DEL BARRIO CASADO - T11, T12
ALEJANDRO MARTINEZ ALEGRE - M21, M22
ROGER MAYNOU GIL - T11, T12
JOSE LUIS TAMARIT MUR - M21, M22

Segon quadrimestre:
MARIA DEL BARRIO CASADO - M31, M32
JOSE IGNACIO ESEBERRI PIEDRA - T11, T12
PEDRO RUFES MARTINEZ - M31, M32
JOSE LUIS TAMARIT MUR - T11, T12

DEGREE COMPETENCES TO WHICH THE SUBJECT CONTRIBUTES

Specific:
CEI-07. Understand applied thermodynamics and heat transfer, their basic principles and their application to engineering problems.

Transversal:
07 AAT N2. SELF-DIRECTED LEARNING - Level 2: Completing set tasks based on the guidelines set by lecturers. Devoting the time needed to complete each task, including personal contributions and expanding on the recommended information sources.

TEACHING METHODOLOGY

Theory (2 sessions per week, 3 ECTS): the teacher presents the fundamental concepts and some demonstrations, complementing key examples and discussion of some applications.
Problems and Guided activities (2 sessions per week, 3 ECTS): the teacher presents representative solving problems; students review the basic concepts and solve some problems under the supervision of the teacher. In efforts to consolidate the concepts students and their magnitudes

LEARNING OBJECTIVES OF THE SUBJECT

After completing the course the student must be able to:
- Know the basic concepts and principles explicitly and understand reasonably the thermal phenomena.
- Feeling comfortable in addressing particular problems in the field of materials engineering.
- Expressing magnitudes in SI units and conversion factors known to other systems of units
STUDY LOAD

<table>
<thead>
<tr>
<th>Type</th>
<th>Hours</th>
<th>Percentage</th>
</tr>
</thead>
<tbody>
<tr>
<td>Hours small group</td>
<td>7,5</td>
<td>5.00</td>
</tr>
<tr>
<td>Self study</td>
<td>90,0</td>
<td>60.00</td>
</tr>
<tr>
<td>Hours large group</td>
<td>52.5</td>
<td>35.00</td>
</tr>
</tbody>
</table>

Total learning time: 150 h

CONTENTS

**Unit 1: Fundamental Concepts**

**Description:**
Unit 1. Fundamental concepts
Introduction to thermodynamics. thermodynamic system, thermodynamic variable, state of equilibrium thermodynamic transformation. Zero Temperature principle. Thermometers and empirical thermometric scales

**Specific objectives:**
To know the basic terms of the thermodynamics

**Full-or-part-time:** 4h
Theory classes: 4h

**Unit 2: Single and Simple Systems**

**Description:**

**Specific objectives:**
To know the fundamental behaviour of the thermodynamic systems

**Related activities:**
Laboratory works

**Full-or-part-time:** 10h
Theory classes: 10h

**Unit 3. Calorimetry and Heat Propagation**

**Description:**

**Specific objectives:**
To know the fundamental concepts of heat and its propagation

**Related activities:**
Laboratory works

**Full-or-part-time:** 9h
Theory classes: 9h
First law of thermodynamics

**Description:**
Expansion work in simple PVT systems. Dissipative work. Conjugate variables and configuration work on other simple systems: surface work, work torsion work electric and magnetic polarization. First Law of thermodynamics. Internal energy. Enthalpy

**Specific objectives:**
To know the 1st law of thermodynamics

**Related activities:**
Laboratory works

**Full-or-part-time:** 6h
Theory classes: 6h

Unit 5: First law of thermodynamics. energy properties and applications

**Description:**

**Specific objectives:**
Know how to apply the 1st law of thermodynamics

**Full-or-part-time:** 7h
Theory classes: 7h

Unit 6: Second law of thermodynamics: Heat Engines

**Description:**

**Specific objectives:**
To know the basic operation of the heat engines and its relation with the 2nd law of thermodynamics

**Related activities:**
Laboratory works

**Full-or-part-time:** 7h
Theory classes: 7h

Unit 7: Second law of thermodynamics: Entropy

**Description:**

**Specific objectives:**
To know the 2nd law of thermodynamics and its entropic statement

**Full-or-part-time:** 8h
Theory classes: 8h
Unit 8: Thermodynamic Potentials

Description:

Specific objectives:
To know the thermodynamic potentials for simple systems

Full-or-part-time: 9h
Theory classes: 9h

GRADING SYSTEM

The final grade for each student is calculated by a weighted average of the marks obtained in partial and final exams, as well as activities aimed at the laboratory. Detailed below the relative weight of each note in the final:
Half-semester exam (multiple choice questions): 20%
Laboratory works: 15%

According to the EEBE academic regulations, students that failed ordinary assessment and have obtained a mark, N, calculated as 65% of the final exam plus 35% of the half-semester exam, 3<N<5, will have a re-assessment test during the period specified in the academic calendar.

BIBLIOGRAPHY

Basic: