

340038 - FENT-F3029 - Fundamentals of Termical Engineering

Coordinating unit:	340 - EPSEVG - Vilanova i la Geltrú School of Engineering
Teaching unit:	729 - MF - Department of Fluid Mechanics
Academic year:	2019
Degree:	BACHELOR'S DEGREE IN INDUSTRIAL ELECTRONICS AND AUTOMATIC CONTROL ENGINEERING (Syllabus 2009). (Teaching unit Compulsory) BACHELOR'S DEGREE IN ELECTRICAL ENGINEERING (Syllabus 2009). (Teaching unit Compulsory) BACHELOR'S DEGREE IN MECHANICAL ENGINEERING (Syllabus 2009). (Teaching unit Compulsory) BACHELOR'S DEGREE IN INDUSTRIAL DESIGN AND PRODUCT DEVELOPMENT ENGINEERING (Syllabus 2009). (Teaching unit Optional)
ECTS credits:	6
Teaching languages:	Catalan

Teaching staff

Coordinator:	JAUME MIQUEL MASALLES
Others:	JAUME MIQUEL MASALLES DAVID MORENO MAESTRO BLAI BUERA MUÑOZ

Prior skills

Previous knowledge of basic thermodynamics and heat transfer.
Basic previous knowledge of the behavior of fluids.
Integral and differential calculus.

Requirements

340022 - Chemistry
340023 - Physics I
340026 - Advanced calculus

Degree competences to which the subject contributes

Specific:

1. CE7. Knowledge of applied thermodynamics and heat transfer. Basic principles and its application in solving engineering problems.

Transversal:

3. SELF-DIRECTED LEARNING - Level 2: Completing set tasks based on the guidelines set by lecturers. Devoting the time needed to complete each task, including personal contributions and expanding on the recommended information sources.
4. EFFICIENT ORAL AND WRITTEN COMMUNICATION - Level 2. Using strategies for preparing and giving oral presentations. Writing texts and documents whose content is coherent, well structured and free of spelling and grammatical errors.
5. TEAMWORK - Level 2. Contributing to the consolidation of a team by planning targets and working efficiently to favor communication, task assignment and cohesion.

340038 - FENT-F3029 - Fundamentals of Termical Engineering

Teaching methodology

- Lectures and participatory classes, consisting of explanation and development of the theory and, if necessary in the resolution of problems. The material user will be available to the student in the Digital Campus section of the subject.
- Practical lessons in problem-solving, where it will seek the maximum involvement of students through their direct involvement in solving the problems. Students must solve in class / outside of class individually problems that are assigned. In the Digital Campus section of the subject, the student can look up the list of problems before they are done in class.
- Hand in resolved problems by students. Submittals will consist on individual, in class or outside class, of some problems of the list or similar, the student will have in the Digital Campus. This activity will be evaluated. The student feed-back can made from the submission of the revised problems.
- Laboratory and simulation practical classes, made directly by students, guided by the teacher, allowing them to directly observe relevant aspects of the theory. The student can look up the explanatory text of the practices to develop in the Digital Campus. The students will give the teacher a copy of the experimental extracted data. Later, students must make a report of the practices carried out. This report will be evaluated and will be delivered before the date set by the teacher.
- Tutorial classes in group or individual.
- INDIVIDUAL WRITTEN TESTS: The students will make two partial exams (individual written tests) of all the knowledge of theory and problems developed in the subject. In the first partial exam (CP1), the knowledge evaluated will be those developed in the first part of the semester, and will be done towards the middle of the semester. In the second partial (CP2) the knowledge developed in the second part of the semester will be evaluated, and it will be done at the end of the semester on the date marked by the EPSEVG (Final Evaluation period). There will be a Final Control of the subject (CFinal) in the Final Evaluation period. Students with grade of CP1 less than 3.50 can be presented in an optional way to this Final Control, which will replace Partial Control 2 (CP2). There will also be a Global Control (CGlobal) of Reevaluation of all knowledge of theory and problems developed in the subject on the date marked by the EPSEVG (Reevaluation period).

Learning objectives of the subject

When the student finishes the subject, he/she has to be capable of:

- Understanding the principles of applied thermodynamics and heat transfer.
- Knowing the principles of thermal equipment and generators.
- Analyzing and solving problems in the area of thermal engineering.
- Interpreting, analyzing, synthesizing and extracting conclusions of results of measurements and tests.
- Writing texts with the structure adapted to the aims of communication.
- Knowing and putting into practice the dynamics teamwork.
- Carrying out assignments from basic directions given by the teacher.



340038 - FENT-F3029 - Fundamentals of Termical Engineering

Study load

Total learning time: 150h	Hours large group:	52h 30m	35.00%
	Hours medium group:	0h	0.00%
	Hours small group:	7h 30m	5.00%
	Guided activities:	0h	0.00%
	Self study:	90h	60.00%

340038 - FENT-F3029 - Fundamentals of Termical Engineering

Content

<p>(ENG) TEMA 1: CONCEPTES FONAMENTALS DE TERMODINÀMICA. PROPIETATS DE LES SUBSTÀNCIES PURES</p>	<p>Learning time: 32h Theory classes: 12h Laboratory classes: 2h Self study : 18h</p>
<p>Description: 1.1. Concept of thermodynamic system. Classification and examples</p> <p>Related activities: A1. Problems of properties of pure substances and equations of state. A6. First individual written test. A7. Computer practice: Calculation of volumetric properties of pure fluids with cubic equations of state and comparison with MINIREF software.</p> <p>Specific objectives: At the end of this teaching unit, the student must be able to:</p>	
<p>(ENG) TEMA 2: PRIMER I SEGON PRINCIPIS DE LA TERMODINÀMICA</p>	<p>Learning time: 32h Theory classes: 12h Self study : 20h</p>
<p>Description: 2.1. Work of volume change in a reversible process in a closed system.</p> <p>Related activities: A.2. Problems of the first and second principles of thermodynamics. A.6. Second individual written test.</p> <p>Specific objectives: At the end of this teaching unit, the student must be able to:</p>	
<p>(ENG) TEMA 3: PRINCIPIS DE TRANSMISSIÓ DE CALOR. APLICACIONS.</p>	<p>Learning time: 28h Theory classes: 10h Laboratory classes: 1h Self study : 17h</p>
<p>Description: 3.1 Introduction to heat transfer mechanisms.</p> <p>Related activities: A3. Problems of heat transfer principles. A8. Laboratory: Determination of thermal conductivity of an insulating material. A11. Second individual written test.</p> <p>Specific objectives: At the end of this teaching unit, the student must be able to:</p>	

340038 - FENT-F3029 - Fundamentals of Termical Engineering

<p>(ENG) TEMA 4: FONAMENTS DE TERMODINÀMICA TÈCNICA</p>	<p>Learning time: 32h Theory classes: 10h Laboratory classes: 3h Self study : 19h</p>
<p>Description: 4.1. Steam power plants cycles: Rankine cycle. Irreversibilities. Superheat and reheat. Regenerative power cycle.</p> <p>Related activities: A4. Problems of fundamentals of Technical Thermodynamics. A9. Laboratory: Determination of the heat balance and the COP of a heat pump as a function of time. A10. Computer practice: Analysis of the operation of a conventional power plant with the "PROPAGUA" software. A11. Second individual written test</p> <p>Specific objectives: At the end of this teaching unit, the student must be able to:</p>	
<p>(ENG) TEMA 5: INTRODUCCIÓ ALS EQUIPS I GENERADORS TÈRMICS</p>	<p>Learning time: 26h Theory classes: 10h Self study : 16h</p>
<p>Description: 5.1. Heat exchangers: Classification. Overall heat transfer coefficient (U). Energy balance equations. Logarithmic mean temperature difference (LMTD). F graphs for different types of heat exchangers. Calculation Method F-LMTD. 5.2. Fuels and combustion: Classification of fuels. Calorific value of the fuel. Chemical equations of combustion (stoichiometric combustion, combustion with air excess and defect). 5.3. Boilers or steam generators: Classification. Use of the boilers. Mass and energy balance applied to a boiler. Efficiency of a boiler.</p> <p>Related activities: A5. Problems of introduction to equipment and thermal generators. A11. Second individual written test.</p> <p>Specific objectives: At the end of this teaching unit, the student must be able to:</p>	

340038 - FENT-F3029 - Fundamentals of Termical Engineering

Qualification system

The different concepts that make up the continuous assessment are:

- INDIVIDUAL WRITING TESTS (75%)
- DELIVERY OF SOLVED PROBLEMS (12%)
- LABORATORY PRACTICES (13%)

To obtain the final grade of FENT, the following equation of the evaluation will be applied:

$$[1] \text{ Final Note of FENT} = \text{Note CP1} * 0.375 + \text{Note CP2} * 0.375 + \text{Note Delivery Problems} * 0.12 + \text{Practice Note} * 0.13$$

Students who have obtained a grade lower than 3.5 in the Note of CP1, may be presented as an optional to a Final Control (CFinal) instead of CP2. This CFinal will be held on the same day and time as the CP2, within the Final Evaluation Period. The equation of the evaluation, to obtain the final grade of FENT, in this case is:

$$[2] \text{ Final Note of FENT} = \text{Note CFinal} * 0.75 + \text{Note Delivery of Problems} * 0.12 + \text{Practice Note} * 0.13$$

There are no minimum notes in any of the previous evaluative acts at the time of applying equations [1] or [2].

RE-EVALUATION:

The student who has: $3.0 \leq \text{Final Note of FENT} \leq 4.9$, has the right to take the re-evaluation of the subject of FENT. The re-evaluation will consist of a Global Control of theory and problems of the subject that will weigh 75%.

Once the Global Control (CGlobal) of re-evaluation is done, the final grade of Reevaluation will be obtained from the following expression:

$$\text{Final Note Re-evaluation} = \text{Note CGlobal} * 0.75 + \text{Note Problem Delivery} * 0.12 + \text{Practice Note} * 0.13$$

The Final Note of FENT after the re-evaluation will be:

- If the Final Note Re-evaluation is equal to or greater than 5.0: The Final Note of FENT = 5.0
- If the Final Note Re-evaluation is less than 5.0: the highest grade will be taken as the Final Note of FENT between the Final Note Re-evaluation and the Final Note of FENT prior to the re-evaluation.

340038 - FENT-F3029 - Fundamentals of Termical Engineering

Regulations for carrying out activities

- Each of the two individual written exams (Partial Controls) consist of two parts: a theory test (which may constitute up to 30% of the test score) and a certain number of problems (up to 100% completion of the test grade). Both tests have the same evaluative weight (37.5%). A minimum grade of the partial controls is not required.
- The Final Control (CFinal) will consist of two parts: a theory test (which may constitute up to 30% of the test score) and a certain number of problems (up to completing 100% of the test grade)). This test has an evaluative weight of 75%. A minimum grade of the Final Control is not required.
- Deliveries of the resolution of practical problems solved individually will be evaluated following the rubric for the realization of the delivery of resolution of problems, which the student will have in advance. The problems solved must be delivered personally to the teacher or through the Campus Athena, and within the allotted time period.
- The reports of laboratory practices will be evaluated according to the rubric established for the realization of them and that the students will have previously. To have a note of the laboratory practices, it is essential to have done the practices in person and present the corresponding reports with the group with which the laboratory practice was carried out.
- If a student presents deliveries of problems and / or practices, at the end he / she will have a grade of the subject even if he / she has not been submitted to the individual written tests (partial controls or final control).

RE-EVALUATION:

- When the Final Score of FENT is lower than 5.0 but equal to or higher than 3.0, you can opt for the Reevaluation. In this case, the theory and problems contents of CP1 and CP2 are reappraised. In the Reevaluation there will be a Global Control of the subject (CGlobal) and this will have a weight of 75%.
- The Global Control (CGlobal) of the re-evaluation will consist of two parts: a theory test (which may constitute up to 30% of the test score) and a certain number of problems (up to 100% of the grade of the test).

340038 - FENT-F3029 - Fundamentals of Termical Engineering

Bibliography

Basic:

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Moran, Michael J.; Shapiro, Howard N. Principles of engineering thermodynamics. 8th ed. Hoboken, N.J.: John Wiley & Sons, 2015. ISBN 9781118412930.

Çengel, Yunus A.; Ghajar, Afshin J. Transferencia de calor y masa : fundamentos y aplicaciones. 4a ed. México [etc.]: McGraw-Hill, 2011. ISBN 9786071505408.

Llorens, Martín; Miranda Barreras, Ángel Luis. Ingeniería térmica. Barcelona: Marcombo, 2009. ISBN 9788426715319.

Complementary:

Çengel, Yunus A.; Turner, Robert; Cimbala, John M. Fundamentals of Thermal-Fluid Sciences. 3rd ed. New York: McGraw-Hill Higher Education, 2008. ISBN 9780071266314.

Kaminski, Deborah A.; Jensen, Michael K. Introduction to thermal and fluids engineering. New York: John Wiley & Sons, 2005. ISBN 0471268739.

Dutta, Binay K.. Heat transfer: principles and applications. New Delhi: PHI Learning, 2006. ISBN 9788120316256.

Chandra, Ramesh. Refrigeration and air conditioning. New Delhi: PHI Learning, 2010. ISBN 9788120339156.

Others resources:

In computer classrooms of EPSEVG, the computer programs (software) used in the course of FENT are installed.