820011 - CTM - Materials Science and Technology

Coordinating unit: 295 - EEBE - Barcelona East School of Engineering
Teaching unit: 702 - CMEM - Department of Materials Science and Metallurgy
Academic year: 2019
Degree: BACHELOR'S DEGREE IN CHEMICAL ENGINEERING (Syllabus 2009). (Teaching unit Compulsory)
BACHELOR'S DEGREE IN MECHANICAL ENGINEERING (Syllabus 2009). (Teaching unit Compulsory)
BACHELOR'S DEGREE IN INDUSTRIAL ELECTRONICS AND AUTOMATIC CONTROL ENGINEERING
(Syllabus 2009). (Teaching unit Compulsory)
BACHELOR'S DEGREE IN ENERGY ENGINEERING (Syllabus 2009). (Teaching unit Compulsory)
BACHELOR'S DEGREE IN ELECTRICAL ENGINEERING (Syllabus 2009). (Teaching unit Compulsory)
BACHELOR'S DEGREE IN BIOMEDICAL ENGINEERING (Syllabus 2009). (Teaching unit Compulsory)
BACHELOR'S DEGREE IN INDUSTRIAL ELECTRONICS AND AUTOMATIC CONTROL ENGINEERING
(Syllabus 2009). (Teaching unit Compulsory)
BACHELOR'S DEGREE IN ENERGY ENGINEERING (Syllabus 2009). (Teaching unit Compulsory)
BACHELOR'S DEGREE IN BIOMEDICAL ENGINEERING (Syllabus 2009). (Teaching unit Compulsory)
BACHELOR'S DEGREE IN MECHANICAL ENGINEERING (Syllabus 2009). (Teaching unit Compulsory)
BACHELOR'S DEGREE IN ELECTRICAL ENGINEERING (Syllabus 2009). (Teaching unit Compulsory)
ECTS credits: 6
Teaching languages: Catalan, Spanish

Teaching staff
Coordinator: José Antonio Benito Páramo
Others: Primer quadrimestre:
JOSE ANTONIO BENITO PARAMO - M21, M22, M23, M24, M25
FERHUN CEM CANER - M13, M14
SALVADOR DOMINGO DURAN - T11, T12, T13, T14, T21, T22, T23, T24
JORDI JORBA PEIRÓ - M11, M12, M13, M14
ISAAC LOPEZ INSA - T11, T12
MERITXELL MOLMENEU TRIAS - M23, M24
MIQUEL PUNSET FUSTE - T21, T22
XAVIER ANDRES ROMERO PEDRET - T13, T14
JOAN SOLA SARACIBAR - M21, M22, T23, T24

Opening hours
Timetable: See ftp://ftp-urgell.upc.es/materials/inici.htm or Atenea, as indicated by the docent.

Prior skills
Basic chemistry, particularly atomic theory, electronic structure and electrochemistry.
Logarithmic and exponential functions.
Trigonometric functions.
Derivatives, integrals and basic calculations.
Office automation software (spreadsheets and word processors).

Degree competences to which the subject contributes
Specific:
On completion of the course, students should be able to:

· Distinguish between the different structures of materials and relate them with the materials' properties and applications.

Transversal:

1. SELF-DIRECTED LEARNING - Level 1. Completing set tasks within established deadlines. Working with recommended information sources according to the guidelines set by lecturers.

Teaching methodology

The course is divided up as follows:
- 20% face-to-face expository classes (theory)
- 10% face-to-face directed classes (problem solving)
- 10% practical work (laboratory)
- 57% self-directed learning (study)
- 3% exams

Learning objectives of the subject

On completion of the course, students should be able to:
· Distinguish between the different structures of materials and relate them with the materials' properties and applications.
· Understand and apply material-testing standards.

Study load

<table>
<thead>
<tr>
<th>Total learning time: 150h</th>
<th>Hours large group: 45h</th>
<th>30.00%</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Hours medium group: 0h</td>
<td>0.00%</td>
</tr>
<tr>
<td></td>
<td>Hours small group: 15h</td>
<td>10.00%</td>
</tr>
<tr>
<td></td>
<td>Guided activities: 0h</td>
<td>0.00%</td>
</tr>
<tr>
<td></td>
<td>Self study: 90h</td>
<td>60.00%</td>
</tr>
</tbody>
</table>
### Content

<table>
<thead>
<tr>
<th>Atomic Structure, Organisation and Movement</th>
<th>Learning time: 30h</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Description:</strong></td>
<td>Theory classes: 9h</td>
</tr>
<tr>
<td>Chemical bonds and types of materials.</td>
<td>Laboratory classes: 5h</td>
</tr>
<tr>
<td>Crystalline structures and imperfections.</td>
<td>Self study : 16h</td>
</tr>
<tr>
<td>Steady-state and non-steady-state diffusion.</td>
<td></td>
</tr>
<tr>
<td>Plastic deformation mechanisms</td>
<td></td>
</tr>
</tbody>
</table>

**Related activities:**
- Practical 1. Metallographic preparation.
- Practical 3. Grain size measurement.

**Specific objectives:**
- Relate materials’ electronic structures, chemical bonds and general properties to one another.
- Relate crystalline structures and their defects to the general behaviour of families of materials.
- Identify diffusion mechanisms in solid materials, their time dependence and applicable equations.
- Study plastic deformation mechanisms in metal materials, the potential interaction between crystalline network defects and the mechanical behaviour of the material. Infer the limit condition for plastic deformation.
- Practical 1. Learn and practise the method for metallographic preparation of metal materials and identify the goodness of a sample by comparing it to established standards.
- Practical 3. Establish and practise the grain size measurement method for metal materials and establish the order of magnitude.

<table>
<thead>
<tr>
<th>Mechanical Properties</th>
<th>Learning time: 33h</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Description:</strong></td>
<td>Theory classes: 10h</td>
</tr>
<tr>
<td>Elastic and plastic deformation.</td>
<td>Laboratory classes: 5h</td>
</tr>
<tr>
<td>Mechanical tests.</td>
<td>Self study : 18h</td>
</tr>
<tr>
<td>Failure and fracture mechanics.</td>
<td></td>
</tr>
<tr>
<td>Fatigue.</td>
<td></td>
</tr>
<tr>
<td>Creep.</td>
<td></td>
</tr>
</tbody>
</table>

**Related activities:**
- Practical 2. Traction and resilience tests.
- Practical 4. Material hardness tests.

**Specific objectives:**
- For students to acquire the ability to define the relevant properties of materials used in structural applications, the tests used to quantify these properties and the values typical of families of materials.
# Phase and Microstructure Control Diagrams.

<table>
<thead>
<tr>
<th>Description:</th>
<th>Theory classes: 9h</th>
</tr>
</thead>
<tbody>
<tr>
<td>Phase diagrams.</td>
<td>Laboratory classes: 2h 30m</td>
</tr>
<tr>
<td>Fe-C diagram.</td>
<td>Self study: 15h 30m</td>
</tr>
<tr>
<td>Cold work and recrystallisation.</td>
<td></td>
</tr>
<tr>
<td>TTT diagrams.</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Related activities:</th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>Practical 5. Evolution of hardness according to the heat treatment and degree of plastic deformation applied to metal materials.</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Specific objectives:</th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>For students to understand how phase diagrams work and the influence of heat treatment on the properties of materials.</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Corrosion and Degradation</th>
<th>Learning time: 11h</th>
</tr>
</thead>
<tbody>
<tr>
<td>Description:</td>
<td>Theory classes: 3h</td>
</tr>
<tr>
<td>Corrosion in metal materials.</td>
<td>Self study: 8h</td>
</tr>
<tr>
<td>Degradation of polymers and ceramics.</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Specific objectives:</th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>For students to acquire the ability to define the conditions in which materials corrode and degrade, the properties that prevent corrosion and degradation and the most relevant families of materials.</td>
<td></td>
</tr>
</tbody>
</table>
820011 - CTM - Materials Science and Technology

### Physical Properties

**Learning time:** 38h  
Theory classes: 13h  
Laboratory classes: 2h 30m  
Self study: 22h 30m

**Description:**  
Electrical conduction. Semiconductors. Dielectrics. Other electrical properties.  
Thermal properties.  
Magnetic properties.  
Optical properties.

**Related activities:**  

**Specific objectives:**  
For students to acquire the ability to define the properties of materials used in electrical, thermal, magnetic and optical applications, the tests used to quantify these properties and the typical values in specific families of materials.

### Material Selection

**Learning time:** 11h  
Theory classes: 6h  
Self study: 5h

**Description:**  
Ashby diagrams.

**Specific objectives:**  
For students to acquire the ability to identify the properties that define a given application and select the material (or family of materials) that can best provide these properties.

### Qualification system

Partial exam: 30%  
Final exam: 50%  
Laboratory: 15%  
Self-directed learning: 5%

The subject has a reevaluation test. The students will be able to access the re-assessment test that meets the requirements set by the EEBE in its Assessment and Permanence Regulations  

### Regulations for carrying out activities

The use of any electronic equipment with wireless communication capabilities is strictly forbidden in the evaluations.
820011 - CTM - Materials Science and Technology

Bibliography

Basic:


Complementary:

